

Beyond default Mixture Allocation Factor (MAF)settings: how can we increase the applicability and scientific-relevance of mixture risk assessments for metal(loid)s?

<u>Charlotte Nys</u>¹, Karel Viaene¹, Karel Vlaeminck¹, Laura De Donno², Marius Schmitt², Koen Oorts¹, Marnix Vangheluwe¹, Hugo Waeterschoot³, Lara Van De Merckt³, Karel De Schamphelaere^{1,2}

Contact: charlotte.nys@arche-consulting.be ¹ARCHE Consulting, Belgium; ² Ghent University, Belgium; ³ Eurometaux, Belgium.

Introduction & Methodology

MFFD

The European Commission calls in the Chemicals Strategy for Sustainability^a to cover the impact of combined exposure into chemical risk assessments under REACH with a **Mixture Allocation Factor (MAF). For environmental compartment**

Eurometaux

MAF for metal(loid)s

MAF_{ceiling}^b

Concentration addition (CA)-based algorithm, solved until:

$$II_{MAF(ceiling)} = \sum RQ_{MAF(ceiling),i} = 1$$

Quantifying conservatism in MAF

 I) MAF algorithm is CA-based, assumes that all substances have the same mode-of-action and all substances contribute, conservatism quantified by

endpoints a default MAF of 5 has been proposed.

Analysis shows that the implementation of a MAF will have a large impact on the environmental risk assessment of inorganics (including metals and metalloids) in soils and waters and may set safety limits below natural background levels.

However, to our knowledge, a specific MAF for metal(loid)s has not yet been derived, as most existing studies do not consider metal(loid)s.

As part of the Metals Environmental Exposure Data (MEED)research programme, we evaluated:

- Relevance of the proposed default MAF for metal(loid)s.
- **Conservatisms** build in the default MAF for metal(loid)s.



 $HI_{MAF(ceiling)}$ =Hazard Index after MAF RQ_{MAF(ceiling)}= Risk Quotient after MAF RQ_i= Risk Quotient for metal *i*

c= dissolved concentration for metal *i* HC5=5% hazardous concentration for metal *i*

- Focus on inorganic-priority contributing substances (I-PCS^c) with species sensitivity distribution approach in PNEC derivation: HC5 and toxicity data extracted from REACH dossiers
- Applied on Flemish freshwater monitoring database^d years 2010-2024 – Cd, Co, Cu, Ni, Pb & Zn bioavailability corrected.

I-PCS = inorganics most contributing to mixture pressure Ag, As, Ba, Cd, Co, Cr, Cu, Hg, La, Mn, Ni, Pb, Se, V & Zn



Raphidocellis

subcapitata

hybrid species between

subcapitata (©DALL-E)

L. stagnalis & R.

Results & Discussion

Quantifying conservatisms in MAF-calculations

MAF for metal(loid)s

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CA predictions of mixture effects at low effect levels

1.00 • Asellus a quaticus (n=4)

Conservatism of applying CA at the HC5-level



Lymnaea

stagnalis

(e.g., Pb-Zn)



Figure 1. $MAF_{ceiling}$ as a function of number of metals measured in a sample. Only those samples for which the original Hazard Index>1 are shown. Numbers indicate the number of samples.

- For the samples where a mixture risk is predicted, the $MAF_{ceiling}$ increases with increasing number of metal(loids) measured in the mixture, with the 90th percentile of $MAF_{ceiling}$ equal to exactly 6.
- However, 41% of the samples in the Flemish monitoring database^d show no mixture risks, i.e., for these samples the MAF_{ceiling}=1.
- Across all samples (i.e., with and without mixture risks predicted), the 90th percentile of MAF_{ceiling} is equal to 4.6.



Figure 2. Cumulative distribution of Mixture Interaction Factors across species for which data on chronic metal(loid) mixture toxicity to aquatic organisms is available.

Mixture Interaction Factor (MIF)^e degree of conservatism that CA provides relative to observed mixture effects at low effect levels (i.e. 10% mixture effect)

- An extensive literature search of available chronic metal(loid) mixture experiments with aquatic organisms was conducted. 34 studies covering 116 mixture experiments with 24 species were identified.
 MIF was calculated for each experiment and averaged per species.
- MIFs range across species between 0.5 (*Chlorella vulgaris*) and 8.9 (*Tetraedron minimum*) (Fig. 2).



Figure 3. Margin of Safety as a function of the number of metals measured in a sample in the Flemish monitoring database.

Margin of Safety (MoS) ^e degree of conservatism that is associated with applying CA at the HC5-level compared to the theoretically more consistent method of applying CA first at the species-level (EC10)

- MoS increases with the number of metals measured (Fig. 3). This indicates that the conservatism in hazard index-based mixture risk estimation approaches, such as MAFcalculations, increases for more complex mixtures.
- The median MoS across samples is 1.8 (10th 90th percentile: 1.2-2.9).

Overall, this case shows the **protectiveness of the proposed MAF of 5** for mixture risks of metal(loids) in a highly industrialized region, such as Flanders.

Conclusions & Outlook

→ CA is not the perfect model and overestimates metal mixture toxicity at low effect levels on average by 1.3-fold.

Median MoS = 1.8

→ Mixture predictions which apply CA at the HC5-level instead of the EC10-level overestimate mixture risks by 1.8-fold

Based on MAF_{ceiling} calculations for the Flemish freshwater monitoring database, our study indicates that a MAF of 5 is protective against risks of metal(loids) a highly industrialized region. As such, a default MAF could serve as a baseline tier for covering risks of unintentional mixtures.

The conservatism embedded in default MAF calculations - a concentration addition (CA)-based approach - was quantified using the Mixture Interaction Factor (MIF) and the Margin of Safety (MoS):

- The MIF indicates that CA overestimates mixture toxicity of metals by 1.3-fold (median MIF).
- The MoS indicates that applying CA at the HC5-level instead of at the EC10-level results in an additional overestimation of mixture risks by 1.8-fold (median MoS).

Outlook: In a next step, a refined MAF will be derived, by taking into account the conservatisms quantified by the MIF and MoS in the MAF algorithm. The refined MAF can increase the scientific relevance under REACH in higher tiers of mixture risk assessments as it takes mixture evidence into account.

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References: ^a European Commission. 2020. https://environment.ec.europa.eu/strategy/chemicals-strategy_en; ^b Backhaus. 2024. Curr Opin Toxicol 100460; ^cARCHE Consulting. 2023. I-PCS identification and MAF level 2-refinement. Report for Eurometaux; ^dwww.vmm.be; ^e Nys et al. 2018. Environ Toxicol Chem 37: 623-642.

ARCHE Consulting Liefkensstraat 35D 9032 Ghent (Wondelgem) - Belgium

